

Problem 1. (20 points)

- (a) You place a particle with energy  $E = \hbar^2 \beta_{21}^2 / 2ma^2$  in an infinite spherical well. Write down the most general wavefunction,  $\psi(\mathbf{r})$ , for this particle. [Hint: Write it in terms of  $\psi_{nlm}(\mathbf{r})$ .]

In this state  $n=2, l=1, m=-1, 0, \text{ or } 1$ . Thus

$$\psi(\vec{r}) = a \psi_{21-1} + b \psi_{210} + c \psi_{211}$$

where  $|a|^2 + |b|^2 + |c|^2 = 1$ .

- (b) A muonic hydrogen atom is made by replacing the electron in the hydrogen atom with a muon. A muon is identical to an electron (same charge, spin, etc.) but is much more massive,  $m_\mu = 207m_e$ . Calculate the ground state energy in electron volts of the muonic hydrogen atom. Calculate the Bohr radius in meters of the muonic hydrogen atom. [Note: Some of the assumptions we made in calculating the hydrogen atom break down when we include a muon instead of the electron. Ignore these details for this calculation.]

$$E_1 = -\frac{m_\mu}{2\hbar^2} \left( \frac{e^2}{4\pi\epsilon_0} \right)^2 = \frac{m_\mu}{m_e} \left[ -\frac{m_e}{2\hbar^2} \left( \frac{e^2}{4\pi\epsilon_0} \right)^2 \right] = -13.6 \text{ eV} \frac{m_\mu}{m_e}$$

So  $E_1 = -2815 \text{ eV} = -2.82 \text{ keV}$

$$a_0 = \frac{4\pi\epsilon_0 \hbar^2}{m_\mu e^2} = \frac{m_e}{m_\mu} \left[ \frac{4\pi\epsilon_0 \hbar^2}{m_e e^2} \right] = \frac{m_e}{m_\mu} (0.529 \times 10^{-10} \text{ m}) = \underline{2.55 \times 10^{-13} \text{ m}}$$

- (c) Calculate the uncertainty relation  $\sigma_z \sigma_{L_z}$ .

$$\sigma_z \sigma_{L_z} \geq \left| \frac{1}{2i} \langle [\hat{z}, \hat{L}_z] \rangle \right|. \quad \hat{L}_z = \hat{x} \hat{p}_y - \hat{y} \hat{p}_x.$$

Thus  $[\hat{z}, \hat{L}_z] = 0 \Rightarrow \sigma_z \sigma_{L_z} \geq 0$

- (d) Calculate the uncertainty relation  $\sigma_x \sigma_{L_z}$ .

$$[\hat{x}, \hat{L}_z] = [\hat{x}, \hat{x} \hat{p}_y - \hat{y} \hat{p}_x] = -\hat{y} [\hat{x}, \hat{p}_x] = -i\hbar \hat{y}.$$

So  $\sigma_x \sigma_{L_z} \geq \frac{\hbar}{2} |\langle \hat{y} \rangle|$

**Problem 2.** (20 points)

A hydrogen atom is in the state  $n = 3$ ,  $\ell = 2$ ,  $m = 1$ . [Note: There is a question on the back of this page.]

- (a) [6 points] Write down the wave function for this hydrogen atom. Express your answer in terms of  $r$ ,  $\theta$ ,  $\phi$ , and constants, as appropriate.

$$\begin{aligned} \psi_{321}(\vec{r}) &= R_{32}(r) Y_2^1(\theta, \phi) \\ &= \frac{4}{81\sqrt{30}} a_0^{-3/2} \left(\frac{r}{a_0}\right)^2 e^{-r/3a_0} \left[ -\sqrt{\frac{15}{8\pi}} \sin\theta \cos\theta e^{i\phi} \right] \end{aligned}$$

s.o.  $\psi_{321}(\vec{r}) = -\frac{a_0^{-3/2}}{81\sqrt{\pi}} \left(\frac{r}{a_0}\right)^2 e^{-r/3a_0} \sin\theta \cos\theta e^{i\phi}$

- (b) [8 points] Find the expectation value of  $r$  for this hydrogen atom.

$$\begin{aligned} \langle r \rangle &= \int r |\psi_{321}|^2 d^3r = \int_0^\infty r |R_{32}|^2 r^2 dr \\ &= \frac{2^4}{3^8 30} a_0^{-3} \frac{1}{a_0^4} \int_0^\infty r^3 r^4 e^{-2r/3a_0} dr \\ &= \frac{2^4}{3^8 30} \frac{1}{a_0^7} 7! \left(\frac{3a_0}{2}\right)^8 = \frac{2^4 \cdot 7 \cdot 6 \cdot 5 \cdot 4 \cdot 3 \cdot 2 a_0}{7 \cdot 2 \cdot 3 \cdot 2} \end{aligned}$$

$$\langle r \rangle = \frac{21}{2} a_0$$

Problem 2 continued:

- (c) [6 points] If we could measure the observable  $\hat{L}_x^2 + \hat{L}_y^2$  for this hydrogen atom what value (or values) could we find and with what probability of each?

$$\hat{L}^2 \psi_{321} = 2 \cdot 3 \hbar^2 \psi_{321} = 6 \hbar^2 \psi_{321}$$

$$\hat{L}^2 = \hat{L}_x^2 + \hat{L}_y^2 + \hat{L}_z^2$$

$$\begin{aligned} \Rightarrow (\hat{L}_x^2 + \hat{L}_y^2) \psi_{321} &= (\hat{L}^2 - \hat{L}_z^2) \psi_{321} = (6 \hbar^2 - \hbar^2) \psi_{321} \\ &= 5 \hbar^2 \psi_{321}. \end{aligned}$$

We measure  $5 \hbar^2$  with 100% prob

**Problem 3.** (20 points)

A hydrogen atom contains two spin 1/2 particles. As discussed in class the total spin of the hydrogen can be either 0 (singlet state) or 1 (triplet state) when the atom is in the ground state. Now consider the case when the hydrogen atom is in the  $\ell = 2$  state. Here we will look at the total angular momentum  $\hat{\mathbf{J}} = \hat{\mathbf{L}} + \hat{\mathbf{S}}$ . Consider the case where the spins of the hydrogen atom are found to be in the triplet state and a measurement of  $\hat{J}_z$  tells us that  $m_j = 0$ . [Note: There are questions on the back of this page.]

(a) [4 points] What are the possible values for the total angular momentum quantum number  $j$ ?

$s=1, \ell=2$  so  $j = 1, 2, 3$

(b) [8 points] For each of the possible values for  $j$  write the state  $|jm_j\rangle$  in terms of the individual states  $|\ell m_\ell\rangle |s m_s\rangle$ .

2x1

$$\begin{aligned} |10\rangle &= \sqrt{\frac{2}{10}} |21\rangle |1-1\rangle - \sqrt{\frac{2}{5}} |20\rangle |10\rangle + \sqrt{\frac{3}{10}} |2-1\rangle |11\rangle \\ |20\rangle &= \sqrt{\frac{1}{2}} |21\rangle |1-1\rangle + \sqrt{\frac{1}{2}} |2-1\rangle |11\rangle \\ |30\rangle &= \sqrt{\frac{1}{5}} |21\rangle |1-1\rangle + \sqrt{\frac{3}{5}} |20\rangle |10\rangle + \sqrt{\frac{1}{5}} |2-1\rangle |11\rangle \end{aligned}$$

Problem 3 continued:

- (c) [4 points] If  $j = 1$  what is the probability of measuring  $m_s = 1$ ? What is the probability of measuring  $m_\ell = 0$ ?

$$\begin{array}{l} m_s = 1 \quad \text{prob } 3/10 \\ m_\ell = 0 \quad \text{prob } 2/5 \end{array}$$

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- (d) [4 points] If we do not measure  $j$  then it can have any value from part (a). In this case what is the total probability that a measurement will give  $m_\ell = 0$ ? [Note: Assume each value of  $j$  is equally probable.]

$$\psi = \frac{1}{\sqrt{3}} (|110\rangle + |20\rangle + |30\rangle)$$

so  $m_\ell = 0$  with prob  $\frac{1}{\sqrt{3}} \left( \frac{2}{5} + \frac{2}{5} \right) = \underline{\frac{1}{3}}$